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may not: a. speed up a reaction b. be solids c. form new substances in a reaction d. be present in living tissues 2.

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Thermochemistry (chapter 5)

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SCH4U Course Notes - earlchemistry

Use the thermochemical equations shown below to

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determine the enthalpy for the reaction: (5 marks) $\text{C}_3\text{H}_8 (\text{g}) + 5\text{O}_2 (\text{g}) \rightarrow 3\text{CO}_2 + 4\text{H}_2\text{O} (\text{l})$ HE =. Microsoft Word - SCH4U - Unit 3 - Version C.doc. $\text{CO}_2 (\text{g}) + \text{C} (\text{graphite}) + \text{O}_2 (\text{g}) \rightarrow \text{CO}_2 (\text{g})$ $\Delta H^\circ = 221.6 \text{ kJ}$. $\text{H}_2 (\text{g}) + \frac{1}{2}\text{O}_2 (\text{g}) \rightarrow \text{H}_2\text{O} (\text{l})$ $\Delta H^\circ = -1160.3 \text{ kJ}$.

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