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Note Taking Guide: Episode 304 Name Aditi Acharya Quantum Numbers N, l, m, s Used to describe an electron in an atom n Principle quantum number Represents main energy level of electron maximum # of electrons in an Energy level = 2n^2 Example: What is the maximum number of electrons that can be in the 5th main energy level? l The 2 nd quantum number • Describes the orbital shape within an ...

note_ep304aditi.pdf - orientation of orbital in space s ...

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Note Taking Guide: Episode 102 Name: Nayan Jacquet International System (SI) Measuring system used in science AKA metric system Length - distance between two points Base unit- meter (M) Instrument- meter stick or metric ruler Mass -amount of matter in an object. Base unit- gram (G) Instrument- balance Volume -amount of space an object occupies Base unit-Liter (L) Instrument- metric ruler or graduated cylinder Time - interval between two occurrences Base unit- seconds (S) Instrument ...

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Note Taking Guide: Episode 502 Oxygen: symbol protons electron distribution Name atomic number electrons Oxygen has valence electrons. Electron Dot Diagram- atom's surrounded by to represent its electrons example electron dot diagrams: Problem Set 1: Li Lewis Structure: diagram representing the arrangement Of electrons in a - valence electrons to become stable.

Pearland Independent School District / Homepage

Note Taking Guide: Episode 301 Name CHEMISTRY: A Study of Matter © 2004, GPB 3.1 Model: A familiar idea used to explain unfamiliar facts observed in nature. Theory: An explanation of facts and phenomena.

Note Taking Guide Episode 301

electronegativity. property that tells how strong an atom's attraction is for an electron. since oxygen has a higher electronegativity than hydrogen, oxygen holds onto shared electrons closer, giving the oxygen a partial negative charge and the hydrogen a partial positive charge.

Note Taking Guide: Episode 501 Flashcards | Quizlet

Note Taking Guide: Episode 701 Name_____ CHEMISTRY: A Study of Matter © 2004, GPB 7.1 Lab results: 1 doz grains of rice = _____g (Use this fact as a conversion

Note Taking Guide: Episode 701 Name

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